



RESULTS OF PRODUCTION AND TESTING OF AFMT-1 BRAND CORROSION INHIBITOR SOLUBLE IN PETROLEUM PRODUCTS.

**(PhD) Khalilov
Jamshid Akmal ugli
Adilova Bahora
Asliddin qizi**

Associate Professor of Karshi University of
Economics and Pedagogy
A student of Karshi University of Economics and
Pedagogy

Annotatsiya

The article examines the physical and chemical properties of corrosion inhibitors obtained on the basis of the processing of organochlorine waste for the oil and gas industry. Corrosion inhibitors of metals were obtained as a result of the synthesis and their level of protection was checked. The results of IR and NMR spectra were studied.

Kalit soʻz

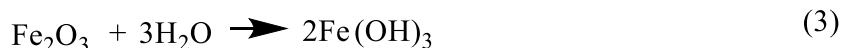
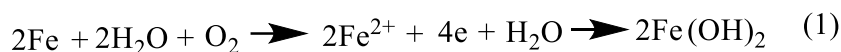
Corrosion inhibitors, nitrogen, organic compounds, fatty acids, gas-condensate well, temperature, mass, level of protection.

Compounds, hydrocarbon-soluble inhibitors account for about 30% of the volume, the largest part of which (~70%) is used in oil refining. Corrosion of these metal-based materials has a large economic impact. According to a recent study by NACE, the total economic loss to the environment due to corrosion is USD 2.5 trillion, which is equivalent to 3.4% of world GDP. [1].

In most cases, the recommended inhibitors are organic compounds of various classes containing heteroatoms: nitrogen, sulfur, oxygen, and phosphorus. The effectiveness of the inhibitory effect of substances increases in the series of heteroatoms: O ^ N ^ S ^ P. However, since the toxicity of products also increases in this series, nitrogen-containing compounds are usually chosen for industrial use. Although it is less effective than compounds containing sulfur or phosphorus, they are a less toxic compound [2].

Corrosion inhibitors are chemicals that are injected into the well in various ways to protect the casing from internal corrosion caused by the produced fluid. It should be noted that some operators further protect parts of upstream structures after the wellhead by choosing the appropriate type and dosage of inhibitors injected into the wells.

The main metal rusting properties are:



Material and Method

Our researched AFMT-1 brand corrosion inhibitor was tested by gravimetric method. This method is used to determine the corrosion rate for corrosion control purposes and to evaluate the protective ability of corrosion inhibitors. The gravimetric method is based on measuring the difference in the mass of control metal samples before and after exposure to a corrosive environment. A limitation of the use of this method is that it characterizes the average corrosion rate without taking into account the unevenness of the corrosion.

In general, when working, it is necessary to follow the current standard GOST 9.506-87 "Methods for determining the protective ability of metal corrosion inhibitors in water-oil environment".

According to it, the product based on amino compounds and fatty acids obtained from the treatment of organochlorine waste is first put into a three-necked flask equipped with a reflux condenser, a thermometer and a stirrer for interaction, and a homogeneous mass is formed. mix until Stirring was continued at a certain temperature for several hours. The obtained corrosion inhibitor was dissolved in gasoline, condensate, and motor oil media at concentrations of 1%, 3%, and 5%. Many studies have been conducted on the resulting solutions.

The physico-chemical properties and analysis results of our PF-1 brand corrosion inhibitor with this synthesized new composition were studied.

Physico-chemical characteristics of AFMT-1 brand corrosion inhibitor obtained on the basis of chlorinated organic waste processing:

table-1.

| Indexes | AFMT-1 |
|--|--------------|
| 1. Appearance | Transparent |
| 2. Color | Pale yellow. |
| 3. Density at 20 °C, g/cm³ | 1...1,3 |
| 4. Nitrogen content, % by weight | 7,0...9, 5 |
| 5. Ph environment at 20 °C | 6,5-7 |
| 6. Level of protection against corrosion at a concentration of 150 mg/l | 98,5 |

| | |
|--|--|
| 7. Solubility: - In gasoline - In the condensate - In the water - In the case of I-20 | Complete Complete 30% of weight gain Complete |
| 8. Fluidity cCt at 20 °C | 15 |

Table-1. Physico-chemical properties of AFMT-1 corrosion inhibitors obtained from chlorinated organic waste processing.

Results and discussion. IR spectrum and analysis of AFMT-1 brand corrosion inhibitor. The IR-spectrum was presented to study the composition and structure of the AFMT-1 corrosion protection inhibitor that we synthesized and used in the test (Fig. 1).

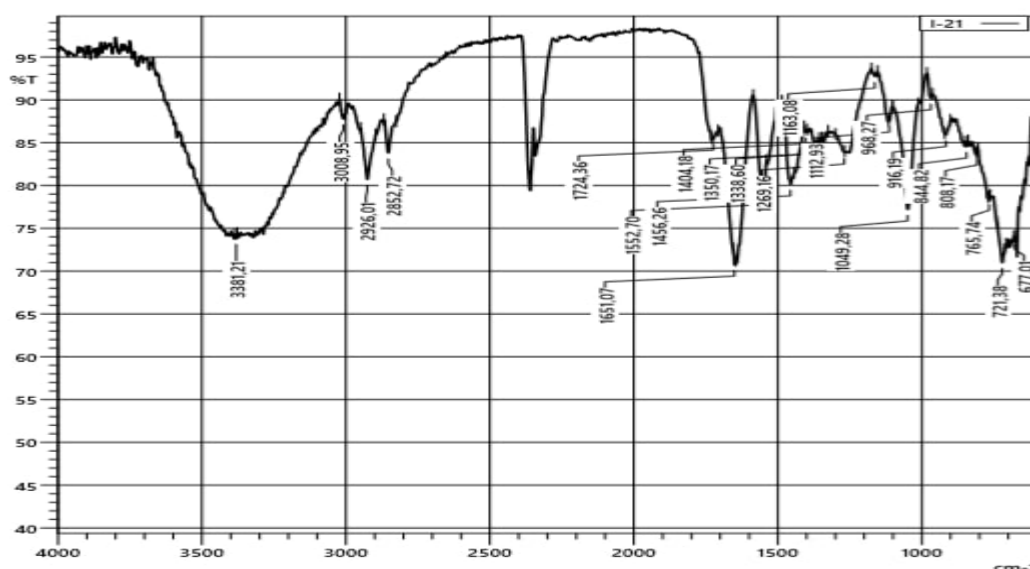


Figure-1. IR spectrum of AFMT-1 brand corrosion inhibitor.

The composition and structure of AFMT-1 corrosion inhibitor was studied using IR-spectrometer technology (IK-Fure, SHIMADZU, Japan) in the range up to 4000 cm^{-1} . The spectrum of the $-\text{N}=\text{C}<$ groups produces valence vibrations in the region of 1651.07 cm^{-1} and in addition 1552.7 cm^{-1} $-\text{NH}_2$ in the structure. $>\text{N}-\text{CH}_2$ in 1350.17 cm^{-1} and valence fields 844.82 – 808.17 cm^{-1} contain absorption lines corresponding to $-\text{CH}_2-\text{CH}_2-$ groups in the aromatic ring.

According to the results of this analysis, our researched corrosion inhibitor contains nitrogen, which shows that it has anti-corrosion properties.

From this table, we can see that the highest level of corrosion protection of the metal surface was applied at a concentration of 6%.

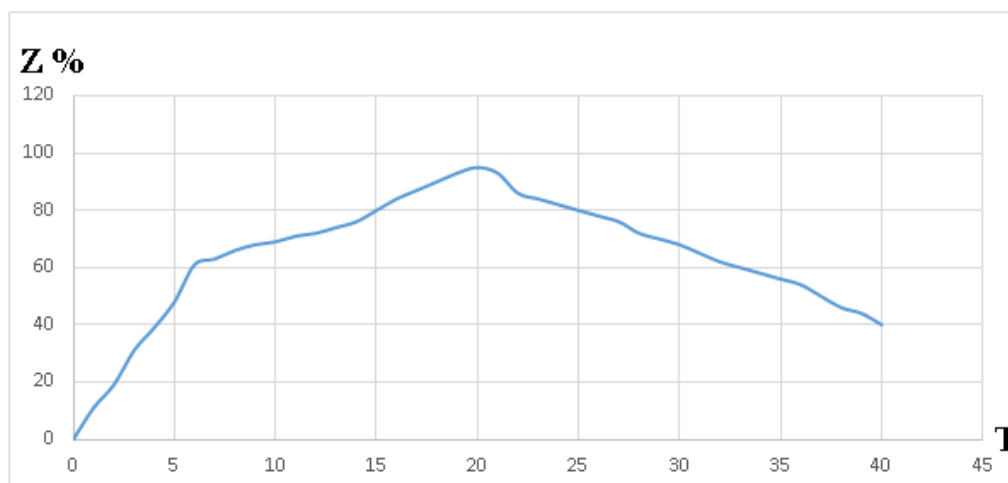


Figure-2. Protection level as a function of temperature.

Figure-2 shows the protection level of protection against corrosion at different temperatures. From this graph, we can see that the optimum temperature for our synthesized AFMT-1 corrosion inhibitor is 20 °C.

table-2

| P:F | Mass ratios | Corrosion rate | Protection level | θ |
|------------|--------------------|-----------------------|-------------------------|----------------------------|
| | 1:1 | 0,065 | 72,31 | 0,7231 |
| | 1:2 | 0,08 | 89 | 0,89 |
| | 1:3 | 0,071 | 78,98 | 0,7898 |
| | 2:1 | 0,058 | 64,5 | 0,645 |
| | 3:1 | 0,051 | 56,7 | 0,567 |

Table 2. Corrosion rates, protection levels and surface coverage coefficient values at different mass ratios of AFMT-1 brand corrosion inhibitor.

As a result of the test research, we can see with the help of table 1 that the best mass ratio of amine compounds and fatty acid is 1:2, and the level of protection in it is 89%.

Conclusion. The physicochemical properties of the AFMT-1 brand corrosion inhibitor synthesized by us and the analysis of the IR and NMR spectrum of the synthesized product were obtained. As a result of the analysis, it was found that this inhibitor contains nitrogen. These compounds have been found to be the most effective against corrosion.

Also, the obtained inhibitor was tested in different environments, at different mass ratios and temperatures. The AFMT-1 brand corrosion inhibitor, obtained as a result of the processing of organochlorine waste, containing nitrogen, was carried out in a condensate medium with a concentration of 1%, 3%, and 6%. As a result of the tests, the level of protection was 83.3, 90.6, 98.6 percent, respectively.

Our researched and tested AFMT-1 corrosion inhibitor can be used in various pipelines in the oil and gas industry in various aggressive environments.

References.

1. Халилов Ж. А. У., Нуркулов Ф. Н., Джалилов А. Т. Синтез и исследование ингибитора коррозии OS-1 для нефтегазовой промышленности

//Universum: технические науки. – 2023. – №. 2-4 (107). – С. 49-53.

2. Jamshid K., Fayzulla N., Abdulahat D. Research And Properties Of AFMD-2 Brand Corrosion Inhibitor For Corrosion Protection Of Oil And Gas Wells //Universum: технические науки. – 2023. – №. 5-8 (110). – С. 12-16.

3. Халилов Ж. А. / Дисс. “Разработка технологии получения ингибиторов коррозии, растворимых в нефтепродуктах, на основе переработки хлорорганических отходов” ст 11-12.

4. Xalilov J.A. Neft quvurlarining korroziyalanishga ta'siri va ularni tadqiq etish // Jurnal “O‘zMU xabarlari” Toshkent — 2022 3/2/1 295-297 bet.

5. История развития и методы совершенствования ингибиторной защиты в ООО «Газпром добыча Оренбург» Д.А. Кузнецов. «Территории Нефтегаз» 2014.

6. Кашковский Р.В.”Перспективы развития метода отдельной оценки вкладов пленки ингибитора и продуктов коррозии в общий защитный эффект”Вестник.